Name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ Date:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

**Year 7 Chemistry Vocab Test**

**Chapter 2.2**

1. What happens when energy is lost from a

substance and the particles slow down and take up less space

2. What causes the particles to vibrate

3. Small particles that all matter is made up of.

4. The attraction between particles

5. Incompressible state of matter where the

particles are strongly bonded and the shape fixed

6. There are no bonds between the particles so it they

can take on the shape of their container

7. Solids liquids and gases

8. What happens when energy is added to a

substance and the particles vibrate faster and take

up more space

9. Particles are able to flow over each other but still

have bonds between them

10. A state of matter rarely found on earth as it is only

found at temperatures above 6000oC